$$\longrightarrow \left(\frac{\partial U}{\partial x}\right)_{s} = -\frac{\left(\frac{\partial S}{\partial x}\right)_{u}}{\left(\frac{\partial S}{\partial y}\right)_{x}} = -T\left(\frac{\partial S}{\partial x}\right)_{u} = 0$$

 $\mathbb{B} \geq \left(\frac{\partial^2 U}{\partial X^2}\right) = ? (>0)$ 

Let  $\left(\frac{\partial U}{\partial x}\right)_s = P$ , then  $P = P(U, X)_s$ 

 $dP = \left(\frac{\partial P}{\partial u}\right)_{x} du + \left(\frac{\partial P}{\partial x}\right)_{y} dx$ 

⇒ divided by dx  
⇒ 
$$\left(\frac{\partial P}{\partial X}\right)_{s} = \left(\frac{\partial P}{\partial U}\right)_{x} \left(\frac{\partial U}{\partial X}\right)_{s} + \left(\frac{\partial P}{\partial X}\right)_{U}$$
  
since  $\left(\frac{\partial U}{\partial X}\right)_{s} \equiv P$    
 $50. \left(\frac{\partial P}{\partial X}\right)_{s} = \left(\frac{\partial^{2} U}{\partial X^{2}}\right)_{s}$   
 $\left(\frac{\partial P}{\partial X}\right)_{U} = \left(\frac{\partial}{\partial X}\left(-\frac{\left(\frac{\partial S}{\partial X}\right)_{U}}{\left(\frac{\partial S}{\partial U}\right)_{X}}\right)\right)$ 

$$= \left(\frac{1}{2\sqrt{2}}\right) \left(-\frac{1}{\sqrt{2}}\right) + \left(\frac{1}{2\sqrt{2}}\right) + \left(\frac{1}{\sqrt{2}}\right) = \frac{1}{\sqrt{2}}\right) \left(\frac{1}{\sqrt{2}}\right) = \frac{1}{\sqrt{2}}$$



Ar example

1.

Equilibrium Condition based U min principle?

↓ rigid, diathermal (impermeable)

2

 $S_{total} = S_1 + S_2 = const.$  $\Rightarrow S_2 = const. - S_1$  Unin

-> isolated

 $\Rightarrow dS_2 = -dS_1 \Rightarrow dU = 0 = T_1 dS_1 - T_2 dS_1$  $= dS_1 (T_1 - T_2) = 0$  $\frac{\Psi}{T_1 = T_2}$ 

2. Legendre Transformation.

Try to involve Intensive Parameters. However, thermodynamic into is incomplete. e.g., Fundamental Eq. Y=Y(X). Intensive parameter.  $P \equiv \frac{\partial Y}{\partial Y}$ it we use Y=Y(P) can we go back to Y=Y(X) -> NO. we will have arbitrary shifs. -> incomplete info. (But."P" is more experimentally relevant)

\* How to avoid such into loss. (while have "P")



 $Y = Y(X) \Leftrightarrow \psi = \psi(P)$ what's the relation between  $(P, \psi) & (X, Y)$ 



$$\Rightarrow P = \frac{Y - \Psi}{X - 0} \Rightarrow \Psi = Y - PX$$

 $Y = Y(x) - P(x) \cdot x = Y(x(P)) - P(x(P)) \cdot x(P)$ 

conversely. if we know  $\Psi = \Psi(P)$  (= Y-PX)

$$Y = Y - PX = Y(P) - P \cdot X(P)$$

 $= \Psi(P(x)) - P(x) \cdot x(P(x))$